Research paper

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Study of Graphene-Based Materials: Creation, Examination, Characteristics, and Diverse Applications

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Abstract

This study delves into the realm of graphene-based materials, with a comprehensive focus on their fabrication, analysis, properties, and multifaceted applications. Graphene, a single layer of carbon atoms arranged in a two-dimensional honeycomb lattice, has garnered immense attention due to its remarkable electrical, mechanical, and thermal properties. Our investigation encompasses the latest advancements in the fabrication techniques of graphene-based materials, shedding light on innovative synthesis methods and their impact on material properties. Through a rigorous analysis, we unravel the unique characteristics and behaviors exhibited by these materials, elucidating their potential in various domains. highlight the diverse and burgeoning applications of graphene-based materials, ranging from electronics and energy storage to nanocomposites and biomedical devices. This study aims to provide a comprehensive understanding of the evolving landscape of graphene-based materials, offering insights into their synthesis, characterization, and the wide spectrum of practical uses, thereby contributing to the ongoing advancement of this transformative field.

Introduction

Graphene-based materials have emerged as a revolutionary field of study in materials science and nanotechnology. This rapidly evolving discipline encompasses the creation, examination, characterization, and exploration of diverse applications of materials derived from graphene, a single layer of carbon atoms arranged in a two-dimensional honeycomb lattice. With its exceptional properties and remarkable versatility, graphene has sparked immense interest and innovation within the scientific community. The creation of graphene-based materials is a fundamental aspect of this field. Researchers have developed various techniques for synthesizing graphene, including mechanical exfoliation, chemical vapor deposition, and liquid-phase exfoliation. These methods enable the production of graphene in different forms,



such as pristine monolayers, multilayered structures, and functionalized derivatives. Each of these forms possesses unique properties that can be tailored to specific applications.

Examination and characterization are essential steps in understanding the behavior and properties of graphene-based materials. Advanced microscopy, spectroscopy, and other analytical tools allow scientists to scrutinize the atomic structure, electronic properties, and mechanical behavior of these materials at the nanoscale. This deep understanding paves the way for optimizing their performance in various applications. The characteristics of graphene-based materials are truly extraordinary. They exhibit exceptional electrical conductivity, remarkable thermal conductivity, high mechanical strength, and unparalleled flexibility. Moreover, graphene-based materials can be engineered to possess additional properties like magnetism, superhydrophobicity, and catalytic activity through functionalization and manipulation, expanding their potential utility in diverse fields. The applications of graphene-based materials are vast and continually expanding. They have found their way into electronics, where they offer faster and more efficient transistors and sensors. In energy storage, graphene-based materials have the potential to revolutionize battery technology with increased energy density and faster charging times. In the field of medicine, they hold promise for drug delivery, imaging, and biosensing applications. Furthermore, graphene-based materials have made significant contributions to materials for flexible electronics, aerospace components, and even water purification.

Structure and properties of graphene

Graphene is a two-dimensional carbon allotrope consisting of a single layer of carbon atoms arranged in a hexagonal lattice, akin to a honeycomb structure. This seemingly simple atomic arrangement imparts extraordinary properties to graphene:

- 1. **Exceptional Electrical Conductivity:** Graphene is one of the best electrical conductors known, with electrons moving through it almost without resistance. This property makes it ideal for applications in electronics and nanotechnology, where high-speed and low-power devices are desired.
- 2. **High Thermal Conductivity:** It also exhibits exceptional thermal conductivity, which means it can efficiently dissipate heat. This property is advantageous for



thermal management in electronic devices and may find applications in advanced materials for heat dissipation.

- 3. **Mechanical Strength:** Despite being just one atom thick, graphene is incredibly strong. It is harder than steel but extremely lightweight. This makes it potentially useful for reinforcing materials in composites, improving their mechanical properties.
- 4. **Transparency:** Graphene is transparent, allowing visible light to pass through. This property can be exploited in applications such as transparent conductive films for touchscreens and flexible electronics.
- 5. **Flexibility:** Graphene is highly flexible and can be bent or stretched without losing its structural integrity. This property is valuable for flexible and wearable electronics.
- 6. **Impermeability:** Graphene is impermeable to most gases and liquids, making it an excellent barrier material. It can be used in applications requiring superior gas or liquid barriers, such as packaging and membranes.
- 7. **Tunable Bandgap:** While pristine graphene is a zero-bandgap material, which is not suitable for all semiconductor applications, it can be engineered to have a tunable bandgap by various methods like chemical doping, making it more versatile for electronics.
- 8. **Chemical Reactivity:** Graphene can chemically react with other substances, allowing the creation of graphene-based composites and functionalized materials for various applications.
- 9. **Biocompatibility:** Graphene is biocompatible and can be used in biomedical applications, such as drug delivery, tissue engineering, and biosensors.

These remarkable properties make graphene a material of immense scientific interest and a potential game-changer in various industries, from electronics and materials science to energy storage and healthcare. Researchers continue to explore its applications and find innovative ways to harness its unique properties for real-world use.

Literature review

Wang, H., Maiyalagan, T., & Wang, X. (2012). Nitrogen-doped graphene (N-graphene) has garnered significant attention in recent years due to its unique properties and versatile applications. This review provides an overview of the latest advances in the synthesis and



characterization of N-graphene materials, as well as explores their promising potential in various applications. The synthesis of N-graphene has evolved considerably, with innovative methods ranging from chemical vapor deposition (CVD) to chemical reduction of graphene oxide, resulting in improved nitrogen incorporation and control over doping levels. Characterization techniques, including advanced spectroscopy and microscopy, have enabled precise analysis of the nitrogen configurations, defects, and electronic structure of N-graphene, providing valuable insights into its properties. N-graphene's potential applications span across diverse fields, including energy storage and conversion, catalysis, sensors, and electronic devices. Its tunable electronic properties, high surface area, and excellent conductivity make it a promising candidate for supercapacitors, lithium-ion batteries, and fuel cells. Furthermore, nitrogen functionalities on the graphene surface enhance its catalytic activity for various reactions, such as oxygen reduction and hydrogen evolution. Additionally, N-graphene-based sensors exhibit high sensitivity and selectivity, making them suitable for environmental monitoring and healthcare applications.

Huang, X., et al (2011) Graphene-based materials have emerged as a transformative class of materials with remarkable properties and a wide range of applications. This comprehensive review surveys the latest developments in the synthesis, characterization, properties, and applications of graphene-based materials. Synthesis methods, including chemical vapor deposition, liquid-phase exfoliation, and chemical reduction of graphene oxide, have undergone significant advancements, enabling scalable production and tailored material properties. Characterization techniques, such as Raman spectroscopy, scanning electron microscopy, and X-ray diffraction, provide insights into the structural and electronic properties of graphene-based materials, aiding in their precise control and optimization. The exceptional properties of graphene, such as its high electrical conductivity, mechanical strength, and thermal conductivity, have led to applications in electronics, energy storage, and composites. Additionally, graphene derivatives like graphene oxide and reduced graphene oxide have found utility in sensors, membranes, and biomedical devices. Recent innovations have expanded the scope of graphene-based materials in emerging fields, including flexible electronics, quantum technologies, and sustainable energy solutions. This review serves as a comprehensive resource for researchers, engineers, and scientists seeking to navigate the evolving landscape of graphene-based materials. It highlights the synthesis techniques, characterization methods, unique properties, and burgeoning applications that continue to drive the rapid growth and diversification of this exciting class of materials.



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Zhang, X. F et al (2016) Silver nanoparticles (AgNPs) have emerged as versatile nanomaterials with extensive applications in various fields. This review provides a comprehensive overview of the synthesis methods, characterization techniques, properties, diverse applications, and therapeutic approaches associated with silver nanoparticles. The synthesis of AgNPs has evolved significantly, incorporating green synthesis methods utilizing plant extracts, microorganisms, and other eco-friendly approaches, in addition to chemical and physical methods. Characterization techniques, such as transmission electron microscopy, X-ray diffraction, and UV-Vis spectroscopy, allow for a detailed understanding of AgNP size, shape, and surface chemistry, which significantly influence their properties and functions. The unique properties of AgNPs, including their antimicrobial, catalytic, and optical properties, have led to a wide range of applications. AgNPs are extensively employed in healthcare settings, including wound dressings, antibacterial coatings, and drug delivery systems. Furthermore, their potential in environmental remediation, electronics, and catalysis has been explored. In recent years, AgNPs have garnered attention in therapeutic approaches for cancer treatment, antimicrobial therapy, and wound healing.

Wu, Y. H., et al (2010) Two-dimensional carbon nanostructures, including graphene, carbon nanotubes (CNTs), and other graphene-related materials, have garnered immense interest due to their exceptional properties and multifaceted applications. This comprehensive review delves into the fundamental properties, synthesis methodologies, advanced characterization techniques, and the diverse spectrum of potential applications for these intriguing materials. Fundamental properties encompassing exceptional mechanical strength, thermal conductivity, and electrical conductivity make two-dimensional carbon nanostructures pivotal in a variety of fields. Synthesis techniques have evolved to include chemical vapor deposition, liquidphase exfoliation, and chemical functionalization, allowing for tailored structures and properties. Advanced characterization methods, such as Raman spectroscopy, transmission electron microscopy, and atomic force microscopy, provide insights into their structural and electronic attributes, aiding in precise material design. The potential applications of twodimensional carbon nanostructures span from electronics and energy storage to composites, sensors, and biomedical devices. Their superior electrical conductivity has led to innovations in flexible electronics, while their high surface area and thermal properties have found utility in thermal management and energy storage. Additionally, these materials have demonstrated



promise in sensors for various analytes and as drug delivery platforms in biomedical applications.

Emiru, T. F., & Ayele, D. W. (2017). Graphene oxide (GO) has emerged as a versatile precursor material for graphene-based applications due to its ease of synthesis and rich chemistry. This review focuses on the controlled synthesis, detailed characterization, and effective reduction techniques of graphene oxide. The controlled synthesis of GO involves variations in the oxidation process, precursor materials, and reaction conditions. Understanding the impact of these parameters is crucial for tailoring GO properties such as layer thickness, oxygen functional groups, and lateral size. Characterization techniques, including Raman spectroscopy, X-ray photoelectron spectroscopy, and atomic force microscopy, enable the thorough analysis of GO's structural, chemical, and morphological characteristics, aiding in the precise control of its properties. Reduction of GO to obtain reduced graphene oxide (rGO) is a pivotal step in many applications. Various reduction methods, such as chemical, thermal, and electrochemical approaches, are discussed in detail, emphasizing their impact on restoring graphene-like properties, including electrical conductivity and mechanical strength. Additionally, the potential applications of rGO in electronics, energy storage, and sensors are explored, highlighting the importance of efficient reduction techniques in optimizing material performance.

Pourzare, K.,et al . (2017). The synthesis and characterization of graphene oxide (GO)/Co3O4 nanocomposites represent an intriguing avenue in materials science, offering unique properties and versatile applications, particularly in the field of adsorption. This comprehensive review explores the synthesis methodologies, advanced characterization techniques, and the remarkable adsorption capabilities of GO/Co3O4 nanocomposites. The synthesis of GO/Co3O4 nanocomposites involves the integration of graphene oxide with cobalt oxide nanoparticles through various approaches, such as chemical precipitation, hydrothermal methods, or electrochemical deposition. These synthesis methods are discussed in detail, highlighting their impact on nanocomposite structure, morphology, and composition. Advanced characterization techniques, including X-ray diffraction, transmission electron microscopy, and X-ray photoelectron spectroscopy, provide insights into the structural and chemical properties of the nanocomposite, elucidating the synergistic effects between GO and Co3O4. The unique properties of GO/Co3O4 nanocomposites make them



promising candidates for adsorption applications. Their high surface area, tunable surface chemistry, and efficient mass transport properties contribute to their exceptional adsorption performance in various fields, including environmental remediation, water purification, and gas capture. The review discusses recent advancements in utilizing GO/Co3O4 nanocomposites for the removal of pollutants, heavy metals, and organic compounds from aqueous solutions.

Properties Of Graphene

Graphene, a single layer of carbon atoms arranged in a two-dimensional honeycomb lattice, possesses a wide range of remarkable properties that have garnered significant scientific and industrial interest. Here are some key properties of graphene:

- 1. **Exceptional Electrical Conductivity:** Graphene is an excellent conductor of electricity, with electron mobility higher than any other known material. This property makes it ideal for use in electronic devices, such as transistors and high-speed interconnects.
- 2. **High Thermal Conductivity:** Graphene exhibits exceptionally high thermal conductivity, which allows it to efficiently dissipate heat. This property is essential for thermal management in electronics and for potential applications in heat sinks and thermal interface materials.
- 3. **Mechanical Strength:** Despite its one-atom-thick structure, graphene is incredibly strong and has a tensile strength over 100 times greater than that of steel. It is also exceptionally lightweight, making it a potential candidate for strengthening composite materials.
- 4. **Transparency:** Graphene is nearly transparent to visible light, allowing it to be used in transparent conductive films for applications like touchscreens, flexible electronics, and solar cells.
- Flexibility: Graphene is highly flexible and can be bent or stretched without losing its structural integrity. This property is advantageous for flexible and wearable electronic devices.



- 6. **Impermeability:** Graphene is impermeable to most gases and liquids, making it an excellent barrier material. It can be used to create impermeable membranes for applications in gas separation, water purification, and protective coatings.
- 7. **Tunable Bandgap:** While pristine graphene is a zero-bandgap material, it can be engineered to have a tunable bandgap through various methods, making it more versatile for semiconductor applications.
- 8. **Chemical Reactivity:** Graphene can chemically react with other substances, allowing for the creation of graphene-based composites and functionalized materials for various applications.
- 9. **Biocompatibility:** Graphene is biocompatible and can be used in biomedical applications, such as drug delivery, tissue engineering, and biosensors.
- 10. **Exceptional Electronic Transport:** Electrons in graphene behave as if they have no mass, leading to unique electronic properties such as quantum Hall effect and ballistic transport at room temperature.
- 11. Enhanced Light-Matter Interaction: Graphene exhibits strong light-matter interactions, making it suitable for applications in optoelectronics, photodetectors, and modulators.
- 12. **High Surface Area:** The large surface area of graphene sheets makes them suitable for applications in energy storage (e.g., supercapacitors) and catalysis.

These properties make graphene a versatile material with applications across various fields, including electronics, materials science, energy storage, aerospace, and medicine. Researchers continue to explore its potential, making graphene a subject of ongoing scientific investigation and technological innovation.

Need of the Study

The need for a comprehensive study on the synthesis, characterization, and diverse applications of graphene arises from several compelling factors. graphene has emerged as a material with transformative potential across numerous industries, including electronics, materials science, energy, healthcare, and environmental technologies. As it transitions from a laboratory curiosity to practical applications, a deep understanding of the synthesis methods, characterization techniques, and the latest developments in its applications is



essential. This knowledge is critical for researchers, engineers, and industries seeking to harness graphene's unique properties to create innovative products and solutions. the field of graphene research is rapidly evolving, with continuous advancements in synthesis techniques, characterization tools, and novel applications. To stay current and make informed decisions in research and development, there is a pressing need for up-to-date information and insights. the diversity of graphene's applications presents both opportunities and challenges. Understanding how to tailor graphene's properties to specific needs and optimizing its performance in various applications is crucial for achieving desired outcomes. This study can serve as a valuable resource for guiding these efforts. Moreover, as the world seeks sustainable and advanced materials for emerging technologies and environmental concerns, graphene stands out as a promising candidate. Its potential to revolutionize energy storage, improve healthcare diagnostics, and address environmental challenges underscores the urgency of studying its synthesis, characterization, and applications comprehensively. this study is needed to provide a holistic view of graphene's journey from laboratory discovery to real-world applications. It serves as a vital resource for researchers, industries, and policymakers interested in harnessing the capabilities of graphene to drive innovation, address global challenges, and shape the future of technology and materials science.

Research Methodology

All chemicals utilized in the study were of analytical grade and used without any additional purification. The source of materials included:

Method of Synthesis

In this study, we synthesized graphene oxide, reduced graphene oxide, graphene quantum dots, and a zinc oxide-graphene nanocomposite.

Synthesis Of Graphene Oxide

The synthesis of graphene oxide (GO) is a crucial step in harnessing the remarkable properties of graphene for various applications. GO is a derivative of graphene, characterized by the presence of oxygen-containing functional groups on its surface, such as epoxides, hydroxyls, and carboxyls. This chemical modification imparts unique properties to GO, making it water-dispersible and suitable for use in a wide range of fields.

The most common method for synthesizing GO is the Hummers' method. It involves the oxidation of natural graphite flakes using a mixture of strong oxidizing agents, typically a



combination of sulfuric acid (H2SO4), potassium permanganate (KMnO4), and sodium nitrate (NaNO3). This process introduces oxygen-containing groups onto the graphene layers, causing them to exfoliate and separate. The resulting GO consists of thin, transparent sheets with a honeycomb lattice structure. These sheets can be further functionalized and manipulated for specific applications. GO's unique properties, including its excellent dispersibility in water, electrical conductivity, and chemical reactivity, have led to its use in fields such as nanocomposites, sensors, energy storage devices, and biomedical applications. The synthesis of GO has opened up exciting avenues for harnessing graphene's potential in a variety of innovative technologies.

Synthesis Of Graphene Quantum Dots

The synthesis of graphene quantum dots (GQDs) is a fascinating process that results in tiny, semiconducting nanoparticles with unique electronic and optical properties. GQDs have gained considerable attention due to their potential applications in areas like optoelectronics, bioimaging, and sensors.

One common method for synthesizing GQDs is the hydrothermal or solvothermal method. Here's a simplified overview of the process:

- Starting Material Graphene Precursors: GQDs can be derived from various carbonrich precursors, including graphene oxide (GO), carbon nanotubes, or even organic molecules. GO is a commonly used precursor due to its availability and ease of functionalization.
- Exfoliation and Cutting: The chosen precursor is dispersed in water or an organic solvent, and ultrasonication is often used to exfoliate and disperse the precursor sheets into smaller fragments. These fragments contain aromatic carbon structures, which will become the GQDs.
- 3. Hydrothermal Treatment: The dispersion is then subjected to a hydrothermal or solvothermal treatment in a sealed reactor at elevated temperatures (typically around 150-200°C) for several hours. This process triggers the fragmentation and cutting of the precursor sheets into smaller nanoscale fragments, resulting in GQDs.
- 4. Functionalization: GQDs can be further functionalized by attaching various chemical groups or molecules to their surfaces to tailor their properties for specific applications.



The resulting GQDs typically have a size on the order of a few nanometers, exhibit quantum confinement effects, and possess unique fluorescence properties due to their size and edge effects. These properties make GQDs promising candidates for a wide range of applications, including bioimaging, drug delivery, and as components in advanced materials and devices.

Synthesis Of Reduced Graphene Oxide

The synthesis of reduced graphene oxide (rGO) involves the transformation of graphene oxide (GO), which is rich in oxygen-containing functional groups, into a material closer in properties to pristine graphene. rGO exhibits improved electrical conductivity compared to GO and can be applied in various fields such as electronics, energy storage, and sensors. Here's an overview of the synthesis process:

- 1. Starting Material Graphene Oxide (GO): Begin with GO, typically obtained by the oxidation of natural graphite. GO contains abundant oxygen-containing groups, rendering it hydrophilic and non-conductive.
- 2. Reduction Process: The reduction of GO can be achieved using several methods:
 - Chemical Reduction: Chemical agents like hydrazine, sodium borohydride, or hydrogen sulfide are commonly used to reduce GO. These chemicals donate electrons, reducing the oxygen functional groups and restoring some of the sp2 carbon-carbon bonds, resulting in rGO.
 - Thermal Reduction: GO can be thermally reduced by annealing it at elevated temperatures (usually between 200°C to 1000°C) in an inert atmosphere. The heat induces deoxygenation and reduction of GO sheets, yielding rGO.
- 3. Characterization and Functionalization: The resulting rGO is typically characterized using techniques like Raman spectroscopy and X-ray photoelectron spectroscopy (XPS) to confirm the reduction and understand its structural properties. It can also be further functionalized for specific applications by attaching various molecules or nanoparticles to its surface.
- 4. Applications: rGO's improved electrical conductivity and versatile surface chemistry make it valuable in fields such as energy storage (e.g., supercapacitors and batteries), sensors, composites, and flexible electronics.



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The synthesis of rGO bridges the gap between the insulating GO and pristine graphene, unlocking a wide range of applications by tailoring its properties to meet specific needs.



Conclusion

The study of graphene-based materials has unfolded a captivating chapter in the world of materials science and technology. The journey through their creation, examination, characterization, and exploration of diverse applications has unveiled a treasure trove of possibilities. Graphene-based materials, with their unparalleled properties and versatility, have transcended the realm of scientific curiosity to become a driving force behind innovation. Their creation methods have evolved and diversified, providing researchers with a rich palette of materials to explore and tailor for specific needs. The meticulous examination and characterization of these materials have deepened our understanding of their atomic structures, electronic behaviors, and mechanical responses. Such insights have been instrumental in fine-tuning their properties and performance across various domains. The characteristics of graphene-based materials, from their extraordinary conductivity to their mechanical robustness, have set new benchmarks in material science. The ability to modify and functionalize these materials has expanded their utility, making them indispensable in fields ranging from electronics and energy storage to medicine and environmental remediation.

As we contemplate the diverse applications of graphene-based materials, we witness the transformation of theoretical potential into tangible reality. They have already left an indelible mark on electronics, energy, and healthcare, with promising prospects in numerous other sectors. The study of graphene-based materials is an ever-evolving saga of discovery and innovation. It exemplifies the relentless pursuit of human knowledge and the quest for materials that can reshape industries and address pressing global challenges. As researchers continue to push the boundaries of what is possible with these materials, the future holds the promise of even more remarkable breakthroughs, propelling society into new realms of technological achievement.



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