Three Component Reaction: An Efficient Synthesis and Reactionsof 3, 4-Dihydropyrimidin-2(1*H*)-Ones Using Baker's yeast in nonaqueos solvent under ultrasonication

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Abstract

Baker yeast (BY) efficient catalyst for organic transformation. A well planned, budget Reducing, eco-friendly deals will created for the Synthesis of 3, 4-dihydropyrimidinenone derivatives from aryl aldehydes, 1,3 diketone and urea in nonaqueous solvent under ultrasonication taking very cheaply, easily available biocatalyst, baker yeast. The bakers' yeast better results in organic solvent, ethanol with no any changes in catalytic power. Different derivatives of benzaldehyde used for this reaction. Synthesize product will be confirmed by using spectral analysis

Keyword- Baker's Yeast, 3, 4-dihydropyrimidin-2(1*H*)-one, 1, 3-dicarbonyl **Introduction**

Multicomponent responses (MCRs) are arising decreasingly as a synthetic plan to develop biologically active compound and natural products.¹ Multicomponent responses are important class in Principles of Green Chemistry and medicine discovery. Also, MCRs have the some merits including coincident creation, Clarity, facile prosecution and envoirmental freidnly.²⁻⁴

The Bigineli reaction is one pot reaction for synthesis of 3, 4 dihydropyrimidinones of from

1.3 diketone, urea and aryl aldehyde.⁵ this multi component reaction for synthesis of 3.4, dihydropyrimidinones by using cheaper biocatalyst baker's yeast.

Dihydropyrimidinones (DHPMs) and their derivatives, products of the Biginelli reaction, are very important pharmacologically active molecules and they have found applications ascalcium channel blockers, a1a-adrenrgic antagonists, antihypertensive agents, inhibitors offatty acid transporters, and in mitotic kinesin inhibition. Also, batzelladine alkaloids containinga dihydropyrimidine core have been found to show potent anti-HIV activity. The most simpleand straightforward procedure for the synthesis of dihydropyrimidinones, originally reportedby Bigineli in 1893, involves the acid-catalyzed one-pot condensation of a, b-keto ester withan aldehyde and urea derivatives. However, the product yields were very Low (20–50%).⁶⁻¹¹ The great biological importance of these heterocyclic compounds has prompted thedevelopment of new improved methodologies for the Biginelli reaction, including transitionmetal Lewis acid catalysis, solid phase synthesis, ionic liquids, activation with certainadditives, microwave-assisted synthesis, ultrasound irradiation, solvent-free techniques, grinding techniques, and many new catalysts.¹²⁻¹⁷

Organic reactions under organic solvent by using biocatalyst we have unexperienced output of the increasing realization of chemical changes. Biocatalytic transformation are more typical reaction with many benefits such as expand selectivity, minimize energy utilization, overcome waste, harmfulness and budget value.¹⁸⁻²⁰

Considering the above fact, we develop an efficiency methodology, for Synthesis of 3,4-

dihydropyrimidin-2(1H)-one derivatives from aldehydes, 1,3-dicarbonyl in nonaqueous solvent in normal condition by taking simple Baker's yeast

Result and Discussion

We describe a systematic and eco- three component reaction using benzaldehyde, urea and ethylacetoacatate in organic solvent under mild condition by employing baker yeast as catalyst. In arrangement of experimental setup, we taking aryl aldehyde (1a) urea and ethylacetoacatateby using baker yeast as traditional standard reaction

We observed that the effect of solvent on yield and time of reaction, standard reaction was proceed in various solvent, Starting of solvent from Starting of solvent from natural solvent i.e. water (H₂O), reaction stir with 20 hrs but reaction not proceed in this solvent because of insolubility of benzaldehyde then we taking organic solvent like ethanol (EtOH), Methanol, Acetonitrile, Dichloromethane, Dimethylformamide, Dimethylsulphoxide, Tetrahydrofuran (THF) (**Table-1 Entry 1-7**). Reaction proceed in all organic solvent but it was interesting

observed that resulting product are high yield in ethanol, ethanol has better yield of product

(4a) and minimum time to achieved goal of reaction (Table 1 entry 2). Therefore ethyl alcohol

was selected for model reaction.

To analyse this reactions the derivatives of aryl aldehyde reacted urea and ethylacetoacatate by using baker's yeast in Ethyl alcohol to get final product in average yields (**Table 2, Entry 1-10**).



Scheme -1

SN	Solvent	Time (h)	Yields ^b (%)
1	Methanol	05	80
2	Ethanol	05	85
3	Acetonitrile	07	70
4	Dichloromethane	07	65
5	Dimethylformamide	07	65
6	Dimethylsulphoxide	08	65
7	Tetrahydrofuran	08	60

Table 1 Screening of solvent on biginelli reaction

^aReaction condition (benzaldehyde 5 mmol, Urea 5 mmol ethyl acetoacetate 5 mmol, 2 gm. baker's yeast and 20 ml solvent at room temperature

^bisolated yield

When urea and ethylacetoacatate was added on aryl aldehyde with electron donating and withdrawing group in high percentage yield of the products (**Table 2, Entry 2-10**).



Table 2 Synthesis of 3, 4-Dihydropyrimidinenone in Ethyl alcohol

Entry	R	Product	Yields (%)	M.P. (⁰ C)
1	Н	4a	85	210
2	4-Cl	4b	80	225
3	3-Cl	4c	80	210
4	$4-NO_2$	4d	75	215
5	3-NO ₂	4e	78	230
6	4-OH	4f	75	245
7	2-OH	4g	65	Not isolated
8	2-Cl	4h	70	200
9	4-N(CH ₃) ₃	4i	85	251
10	4-OMe	4j	80	170

^aReaction condition benzaldehyde (0.5 g), Urea 5 (0.4 g ethyl acetoacetate (0.9 g), 2 gm. Baker yeast and 20 ml solvent at room temperature

^b-Isolated yield

^c Product were characterised by Melting point , ¹H NMR, ¹³C NMR

Bakers' yeast is biocatalyst and it contain many enzymes are present. A base is required to catalyse this reaction and in every enzyme histidine is present which contain imidazole as basic site. The histidine present, might have acting as a base in this transformation. we are proposing herewith plausible mechanism of this reaction.



Conclusion

Final conclusion that First time baker yeast is used to carried out biginelli reaction in ethyl alcohol. The overall synthetic process is green and high yielding. Thus further studies on some preparation of heterocyclic compound is carried out

4.1 General

Most of Chemicals we are used from Alfa Aesar and some chemical are used from sigma-Aldrich. Baker's yeast purchase from native provider.1H NMR and 13C NMR Spectra was characterised in sophisticated analytical instrumentation facility (SAIF), University of Punjabat close tempreture in CDCl3 solvent.

General experimental procedure for Knoevenagel condensation of malononitrile and aldehydes

A mixture of aryl aldehyde (0.5 g), urea (0.4 g) and Ethyl acetoacetate (0.9 g was stirred at ultrasonication in Ethyl alcohol (20 ml), once homogenous solution is created then baker's yeast (2 g) is added in R. B. Flask. Then reaction was proceed at 30 $^{\circ}$ c under ultrasonication , the monitoring of the reaction by T.L.C. method, once a pair of hours reaction composition was filtered out to remove the catalyst and get final product. Final product was purification by using column chromatography.

Spectral characterisation of synthesize compound

3,4-dihydro-4-phenyl-5-propionylpyrimidin-2(1H)-one [4a]

¹H NMR (400 MHz, CDCl3) δ (ppm): 7.14 (d, 1H), 7.21 (dd, 1H) 7.30 (dd, 1H), 7.93(S,1H) 13 CNMR (100 MHz, CDCl₃) δ (ppm): 69.3, 113.7, 126.41, 128.0, and 128.7,135,155

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